

Green Chemistry — Polycarbonate Plastic Production

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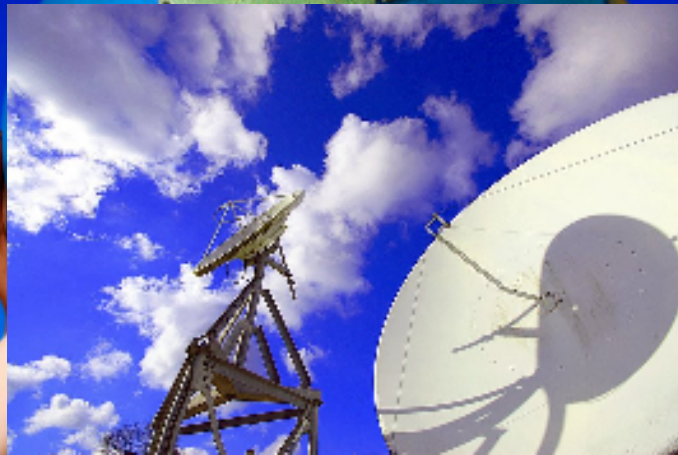
Content

- Properties and uses of polycarbonate
- Traditional method
- Greener method
- Comparison of 2 methods
- Principles of green chemistry In this process

Properties of polycarbonate

- Do not conduct electricity
- high heat resistance
- flame retardant
- Durable
- Strong
- usable in a wider temperature range
- highly transparent
- easier to mould

Electronic components : dielectric ,
electrical and telecommunications
hardware



Construction materials : sound walls



Data storage : Blu-ray Discs



Automotive, aircraft, and security components : bullet-proof "glass"



Niche applications : sunglass

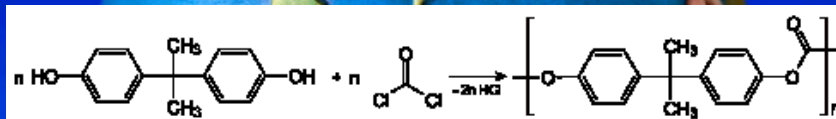


Medical applications

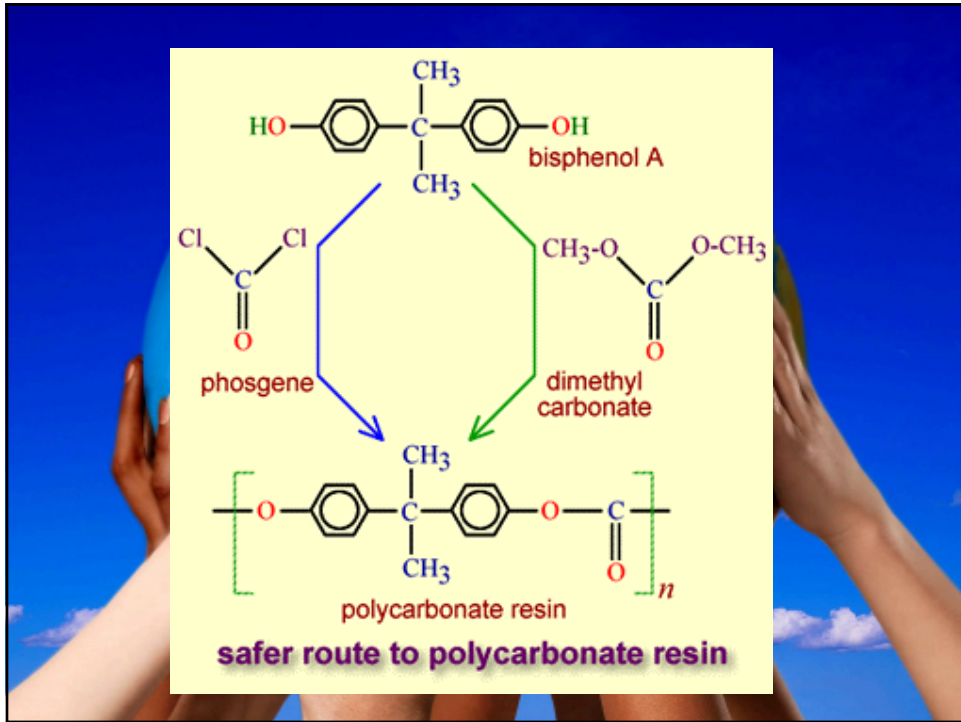


Traditional method

- Bisphenol A + Phosgene \rightarrow Polycarbonate



- Condition : addition of sodium hydroxide (NaOH)
- By-product : hydrochloric acid (HCl)



Greener method

- Bisphenol A + **Dimethyl carbonate** → polycarbonate
- By-product : methanol
(raw material of dimethyl carbonate)

$$\begin{array}{c}
 \text{H} \\
 | \\
 \text{H}-\text{C}-\text{O}-\text{H} \\
 | \\
 \text{H}
 \end{array}$$

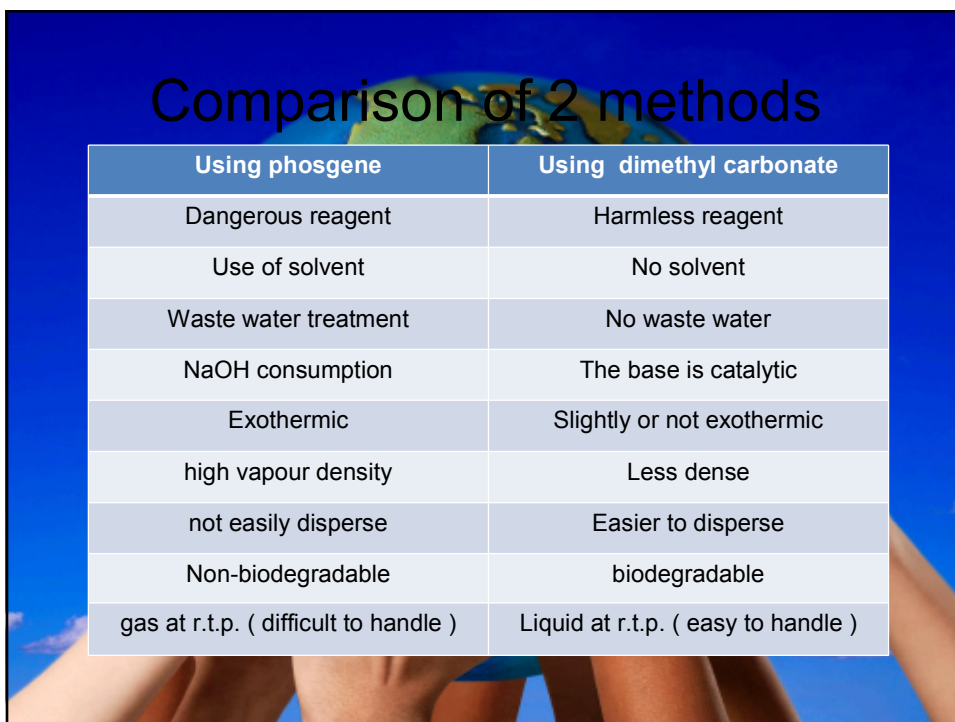
Production of dimethyl carbonate

- carbon monoxide + methanol + oxygen → dimethyl carbonate and water



Comparison of 2 methods

Using phosgene	Using dimethyl carbonate
Dangerous reagent	Harmless reagent
Use of solvent	No solvent
Waste water treatment	No waste water
NaOH consumption	The base is catalytic
Exothermic	Slightly or not exothermic
high vapour density	Less dense
not easily disperse	Easier to disperse
Non-biodegradable	biodegradable
gas at r.t.p. (difficult to handle)	Liquid at r.t.p. (easy to handle)



1. Prevention (prevent waste)

- Traditional method :
Production of phosgene produce CCl_4 as by-product which is undesirable
- Greener method :
the by-product (methanol) can be reused to make dimethyl carbonate

2. Atom economy (minimize by-product)

Comparison between formula mass of by-products :

- Traditional method
$$\%AE = 290.7 / (98.92 + 228.29) \times 100\% = 88.8\%$$
- Greener method
$$\%AE = 290.7 / (90.08 + 228.29) \times 100\% = 91.3\%$$
- Greener method has **higher** %AE.

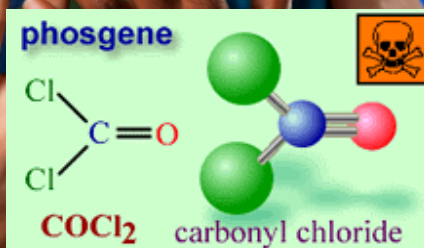
3. Less hazardous synthesis (Little or no toxicity)

- Traditional method :
HCl is strong acid and it is corrosive
- Greener method :
Methanol is relatively harmless

4. Designing safer chemicals

Traditional method :

- phosgene is **highly toxic**
- used as a chemical weapon in the WWI
(responsible for about **80%** of chemical fatalities)





5. Safer auxiliary substances (e.g. solvents, separation agents, etc.)

- Traditional method :
Required organic solvent which is carcinogenic
- Greener method :
Use water as solvent



6. Energy efficiency

- Traditional method :
Require higher temperature to trigger off the reaction
- Greener method :
Need lower temperature to activate the reaction

7. Use of renewable resources

- Traditional method :
 Cl_2 is non-renewable which is the raw material of phosgene
- Greener method :
Methanol, CO and O_2 are renewable which are raw materials of dimethyl carbonate

8. Reducing derivatives (fewer steps of reaction)

- Traditional method :
The purification of CO used in producing phosgene is more complicated and it is difficult to separate the product from by-product
- Greener method
Purification is not required and it is easy to separate the product from methanol (low boiling point)

9. Catalysis

- Traditional method :
Activated carbon is required and it is difficult to produce
- Green method
Copper chloride with 5% KCl additive is needed to produce dimethyl carbonate

10. Design for degradation (decomposition)

- Traditional method :
The excess chemicals are not degradable
- Greener method :
The remaining dimethyl carbonate is biodegradable

11. Use real-time analysis for pollution prevention

- Traditional method :
Phosgene is a colourless gas and has unnoticeable odour, leakage is difficult to be detected
- Greener method :
All chemicals are liquids, leakage is relatively easy to be detected instantly

12. Accident prevention

- Traditional method :
Phosgene is highly toxic, direct contact is fatal so lots of accident prevention measures are needed
- Greener method :
Direct contact of chemicals is unlikely to be fatal

